

Development of a simple system for on-site detection of hydrogen peroxide.

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Background / Objectives

Background

- Hydrogen peroxide (H₂O₂) is mainly used in industrial processes like pulp and textiles bleaching as well as for disinfection.
- Peak exposures are characteristic in such workplaces and are difficult to evaluate with the existing methods.
- Concentrated H₂O₂ is also used as precursor in manufacturing improvised explosive (IE) by criminals.
- In that case, security agencies need miniaturized systems allowing low detection limit for H₂O₂ in order to locate the IE manufacturing sites.

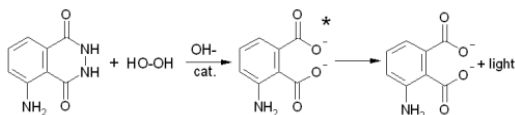
Objectives

- To develop an analytical method based on a portable luminometer allowing to measure H₂O₂ peak exposure (15 minute sampling) with low detection limits;
- To apply it on-site for evaluating occupational exposure and ability to locate IE manufacturing sites.

Methods

Method development

- Off-line procedure based on sampling gaseous H₂O₂ with an impinger filled with water;
- Measurement of the dissolved H₂O₂ with a luminescence-based method using a mixture of horse radish peroxidase (HRP) and luminol, following the reaction scheme:



Luminol reacts with H₂O₂ to produce an electronically excited 3-aminophthalate, which emits in the blue (450 nm).



- Use of a commercially available portable luminometer (Hygienia EnSure, Camarillo, CA, USA).

- Preliminary tests were done in order to optimize the HRP type, ratio HRP/luminol, pH and duration of the recording signal. A bench luminometer (Multiplate reader Infinite M200, Tecan, Männedorf, Switzerland) was used for comparison purpose and considered as reference.

Method validation for the measurement of H₂O₂ in air

- Tests in a cabin (10 m³) were done in order to evaluate the H₂O₂ levels emitted during:
 - colour preparation and application on paper sheets (modelisation of the coloring activities in an hairdresser salon)
 - concentrating a diluted H₂O₂ solution on an heating plate (modelisation of the first step necessary to prepare improvised explosives).
- Field measurements :
 - in an hairdresser salon
 - in the field (ArmaSuisse - Thun)

Acknowledgements

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Results / Discussion

Method validation

- Figure of merit of the luminescence method for H₂O₂ in air.

Reactive mix	HRP type X 0.8 U/ml – Luminol 8.3 mM in 0.1M Tris pH 8.4
Calibration	Polynomial fit (second order)
Limit of detection	0.25 µM impinger 15 ml H ₂ O → 7 µg H ₂ O ₂ /m ³
Repeatability	< 10%
Reading duration	6 minutes

Coloring activities at an hairdresser salon

- Modelisation in a cabin:

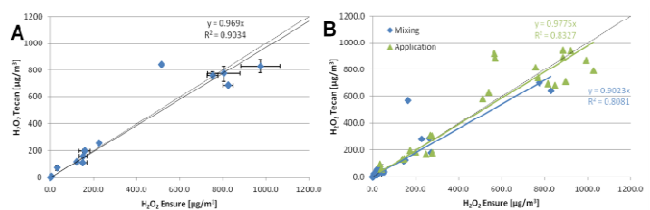


Fig. 1: Comparison of H₂O₂ results determined with the reference luminometer (Tecan) and the portable one (Ensure).

- Cumulative H₂O₂ levels (mixing+application, 15 min duration);
- Contribution of the mixing and the application activities.

- The measured levels of H₂O₂ in the cabin could be very high (above the OEL of 710 mg/m³; Fig. 1).

- Field measurement

Sampling N°	Sampling place	H ₂ O ₂ [µg/m ³]	Activity
1	Fix, next to the client	5 ± 1	Preparation and color application (locks of hairs). Duration : 37 minutes.
2	Fix, next to the hairdresser	24 ± 1	
3	Personnal (hairdresser)	38 ± 3	Preparation and color application (locks of hairs). Duration : 15 minutes.
4	Fix, next to the client	11 ± 1	
5	Fix, next to the hairdresser	42 ± 3	After the coloration Background. Duration : 15 minutes.
6	Personnal (hairdresser)	106 ± 6	
7	Fix, next to the client	6 ± 1	
8	Fix, next to the hairdresser	6 ± 1	

- The H₂O₂ levels were low; the activity of color application on the locks is the most of concern (use of color with higher H₂O₂ content).

IE preparation (concentrating a diluted H₂O₂ solution on an heating plate)

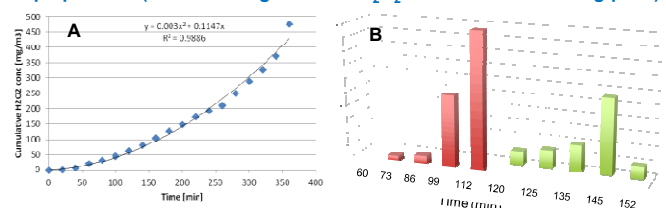


Fig 2: A. Evolution of the concentration of H₂O₂ over time in the closed cabin. B. Evolution of the concentration of H₂O₂ over time in two impingers, positioned at distance <5m (in red) and between 5-20 m (in green) in the field (ArmaSuisse) determined with the portable luminometer.

- H₂O₂ is volatile (Fig. 2 A.) and the developed method is sensible enough to detect emissions from heated solutions at distances of some meters in real situations (Fig. 2 B.).

Conclusion / Perspectives

- Preliminary testing in an occupational simulation has been successfully completed. Successful low-level environmental sampling/detection indicates a feasible method applicable for short term exposure level for H₂O₂ in occupational setting.

- Detection of H₂O₂ from bomb factories producing IE is possible with the developed method and opportunities exist for continuous on-site monitoring via portable set-up systems.